The CH 231 course website is accessed through my.oregonstate.edu

Midterm Exams:  
Exam 1  Thursday  October 18, 2012  7:00-8:20pm  
Exam 2  Thursday  November 15, 2012  7:00-8:20pm

Final Exam:  Final Exam  Tuesday  December 4, 2012  8:00-9:50pm

Lectures:  
MWF  8:00-8:50am  125 LPSC  Nafshun  
MWF  9:00-9:50am  125 LPSC  Nafshun  
MWF  11:00-11:50am  125 LPSC  Watson  
MWF  1:00-1:50pm  125 LPSC  Nyman  
MWF  2:00-2:50pm  125 LPSC  Barth  
MWF  3:00-3:50pm  125 LPSC  Sleszynski

Instructors:  
Dr. Richard Nafshun  Richard.Nafshun@oregonstate.edu  
Office Hours:  W 10am-10:50am (239 LPSC), W 11am-11:50am (239 LPSC)  
Dr. Philip Watson  Philip.Watson@oregonstate.edu  
Office Hours:  M Noon-12:50pm (239 LPSC), W 3:00-3:50pm (147 GILB)  
Dr. May Nyman  May.Nyman@oregonstate.edu  
Office Hours:  M 2:00pm-2:50pm (239 LPSC), W 2:00pm-2:50pm (239 LPSC)  
Dr. Marita Barth  Marita.Barth@oregonstate.edu  
Office Hours:  M 3:00pm-3:50pm (239 LPSC), W 3:00pm-3:50pm (239 LPSC)  
Dr. Neal Sleszynski  Neal.Sleszynski@oregonstate.edu  
Office Hours:  M 4:00pm-4:50pm (239 LPSC), W 4:00pm-4:50pm (239 LPSC)

The above schedule indicates there are ten faculty office hours per week. CH 231 students are welcome to attend any of these for chemistry related questions. For example, a student in Dr. Barth’s lecture section is welcome to attend Dr. Sleszynski’s Monday 4:00pm office hour due to a schedule conflict.

Teaching Assistants (TA):  TA Office Hour are in the Mole-Hole: http://www.chem.orst.edu/courses/molehole.html

Textbook and Supplies:  
Chemistry: A Molecular Approach, by Tro, 2/ed, Text, Prentice Hall (Required)  
Note: This is NOT the Principles of Chemistry: A Molecular Approach text used in CH 121.  
Mastering Chemistry, Online Homework, Prentice Hall (Required)  
www.onlinechemlabs.com, Online Laboratories, Online Labs, LLC (Required)  
Any Scientific Calculator can be used broadly in CH 231, CH 232, and CH 233. Phones are not permitted to be used as calculators and may not be used during assessments.

Course Coverage:  
Matter & Measurement (Chapter 1), Atoms & Elements (2), Molecules (3), Stoichiometry (4), Thermochemistry (6), Quantum Mechanics (7), Periodic Trends (8), Bonding I (9).

A grade in this class will be obtained through the point system outlined on the CH 231 Student Grade Sheet; 552 points for an A, 534 for an A-... If you write a higher percentage score on the final exam than on the midterms exams, the final exam percentage will be used as a score for all three term exams (see the CH 231 Student Grade Sheet). There is nothing to sign-up for. It is an automatic win-win situation. This is a mechanism which gives students who experienced educational difficulty during the term for any reason a chance to demonstrate that they have mastered the material and to profit from that mastery. There are no "make-up exams," "early exams," "alternate exams," "extra-credit assignments," or "deals." If you miss Exam 1 or Exam 2 due to a choice or unavoidable event, the above mechanism will be used. If you were sick during Exam 1 but took the exam, know that you can ace the Final Exam and still earn an "A" in CH 231. If you score 96% on Exam 1 and 100% on Exam 2, know that you need master a small amount of new material to perform excellently on the cumulative final exam to maintain your "A." You must take the CH 231 final exam as scheduled. You cannot reschedule the final exam.

Note: The ACS General Chemistry Exam will be administered as part of CH 233 in Spring 2013.